



## 五、一些形成C-C键的基本反应

### (三) 自由基参与的反应

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## 一、概论

## 二、基础知识

### 构象分析

### 有机反应的热力学和动力学

### 构象对反应活性的影响

### 立体电子效应

## 三、氧化态的调整

### 烯烃、醇和其他化合物的氧化

### 烯烃、羰基化合物和其他化合物的还原

## 四、C-X键形成反应

## 五、一些形成C-C键的基本反应

### 烯醇和烯醇负离子化学

### 有机锂、镁和铜试剂的制备和反应

### 自由基反应

### 烯基化反应

## 六、周环反应

### 非直观Diels-Alder反应

### 1,3-偶极环加成反应

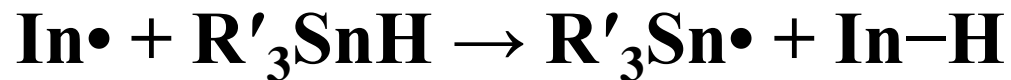
### 电环化反应

### sigmatropic重排

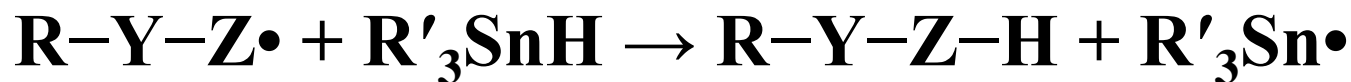
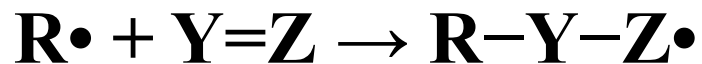
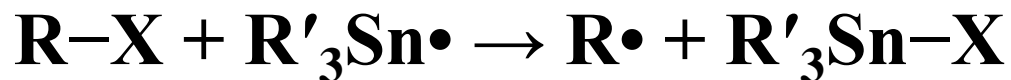
## 七、阳离子参与的C-C键形成反应

# 自由基参与的C-C键形成反应：基本过程

引发



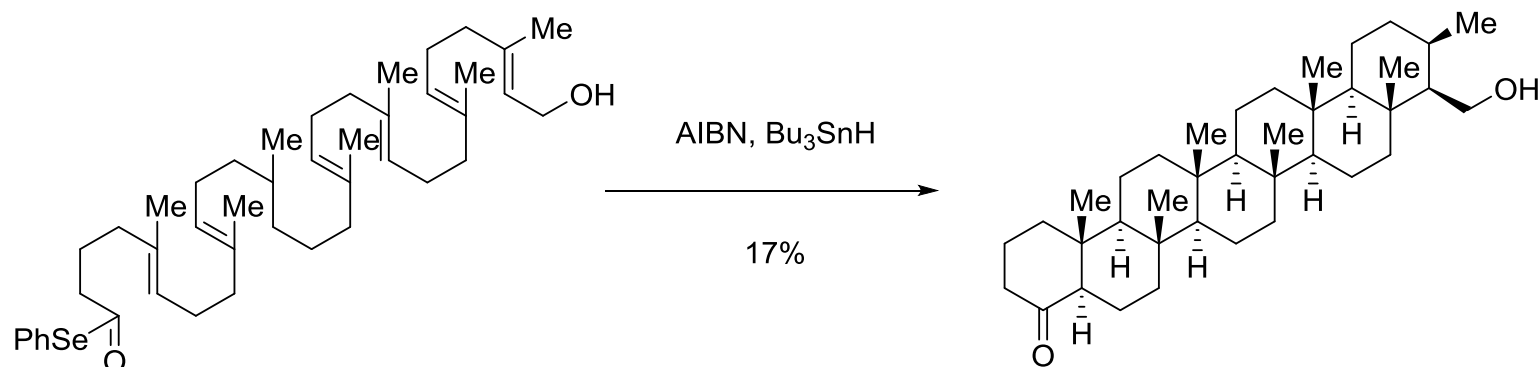
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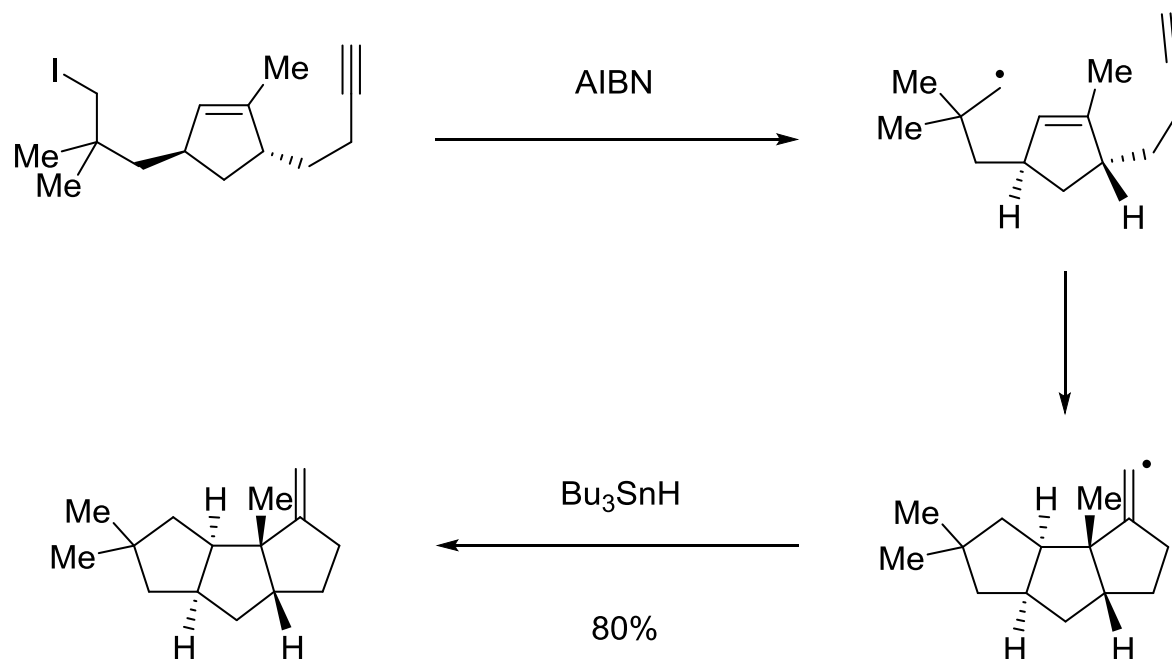
## 自由基参与的C-C键形成反应：优点

- **Neutral reaction conditions**
- **Compatibility with radical acceptors containing functional groups – carbonyls, enol ethers, and enamines**
- **Compatibility with Lewis acids**
- **No necessity for protection of alcohol and amine functional groups**
- **Compatibility with protic solvents – potential for reaction in aqueous systems**
- **Ease of quaternary center formation**

# 自由基的优点：两个经典实例

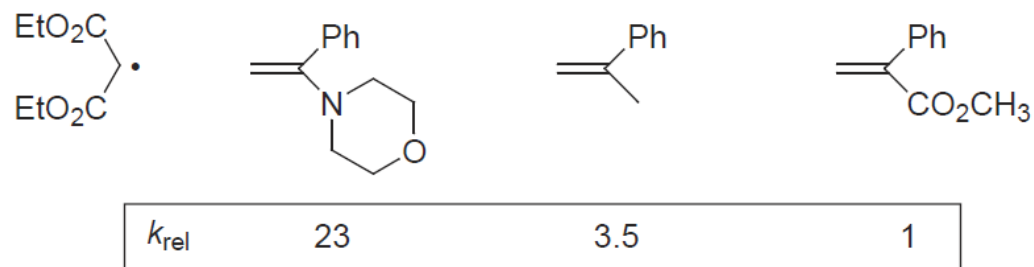
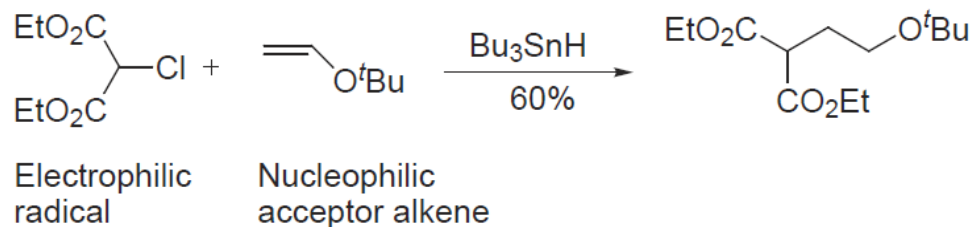
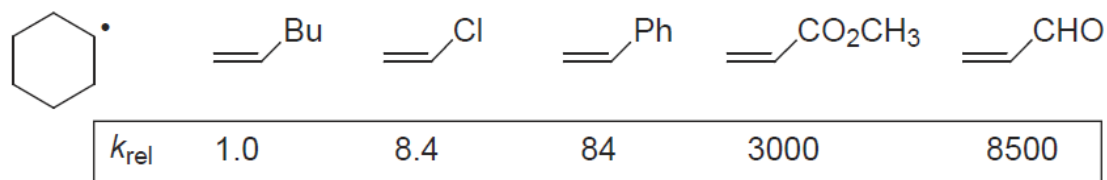
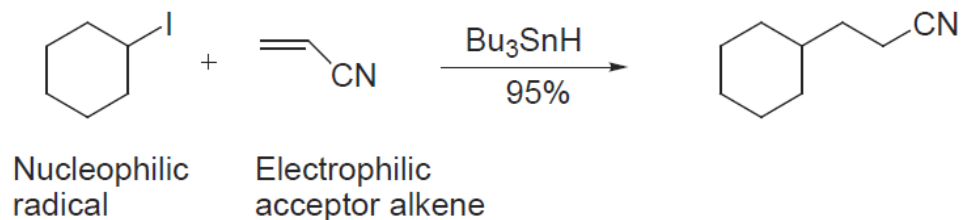


G. Pattenden, et al. *J. Chem. Soc., Perkin Trans. 1*, **1999**, 843.



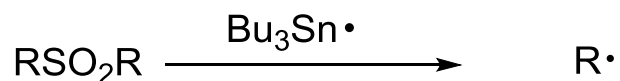
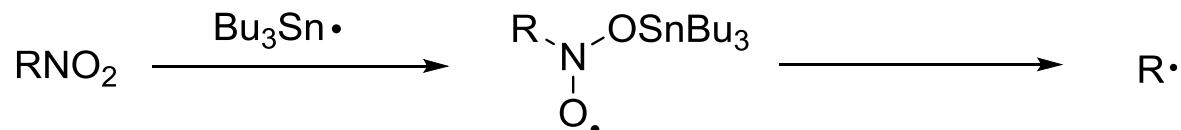
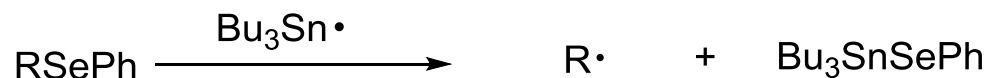
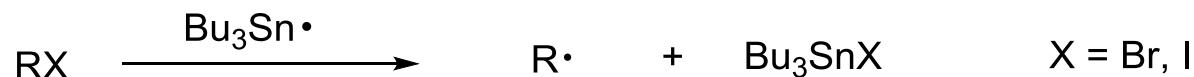
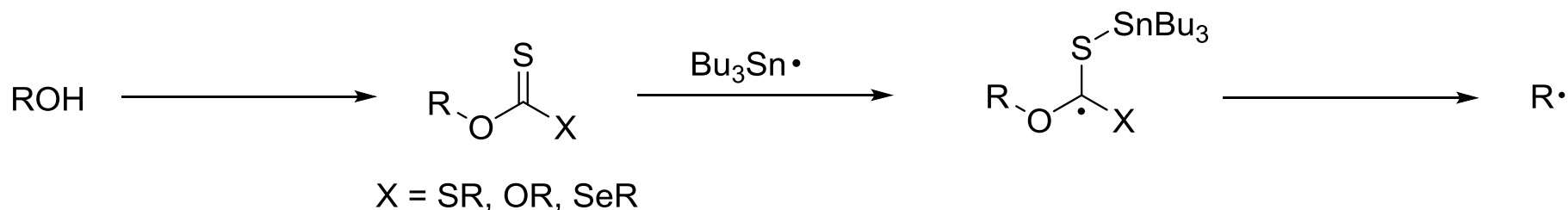
D. P. Curran, et al. *J. Am. Chem. Soc.* **1985**, 107, 1448.

# 自由基参与的C-C键形成反应：电子效应



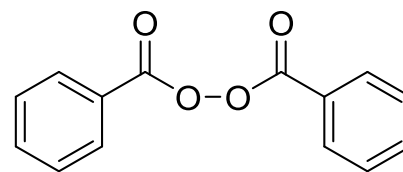
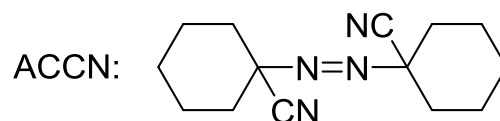
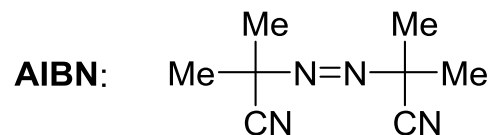
# 自由基参与的C-C键形成反应：引发官能团

Initiator Groups



# 自由基参与的C-C键形成反应：引发剂、氢化物

引发剂



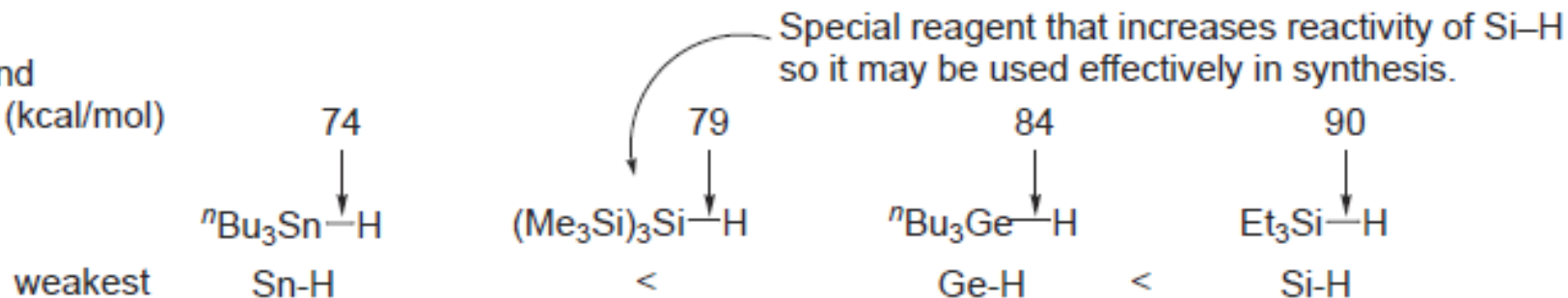
**Et<sub>3</sub>B/O<sub>2</sub>, Et<sub>2</sub>Zn/O<sub>2</sub>, Me<sub>3</sub>Al/O<sub>2</sub>**

hν

氢化物

Ru/Ir complex or organic dye, visible light

M-H Bond  
strength (kcal/mol)



More competitive reduction by H• abstraction from reagent

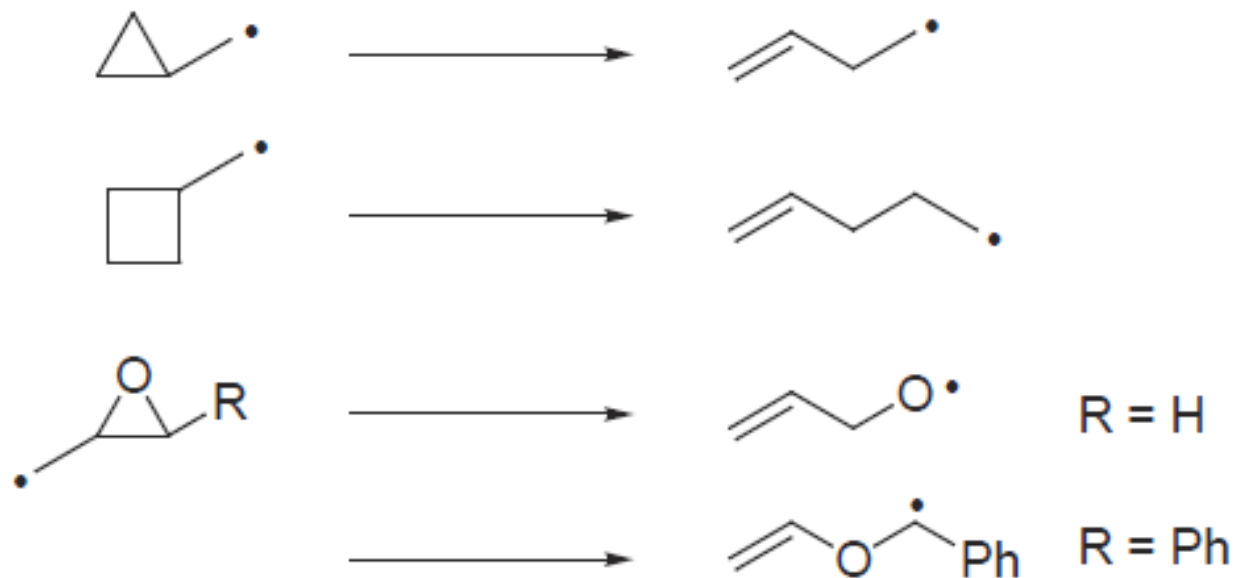
Giese *Tetrahedron Lett.* **1989**, 30, 681.

Ingold *Int. J. Chem. Kinet.* **1969**, 7, 315.



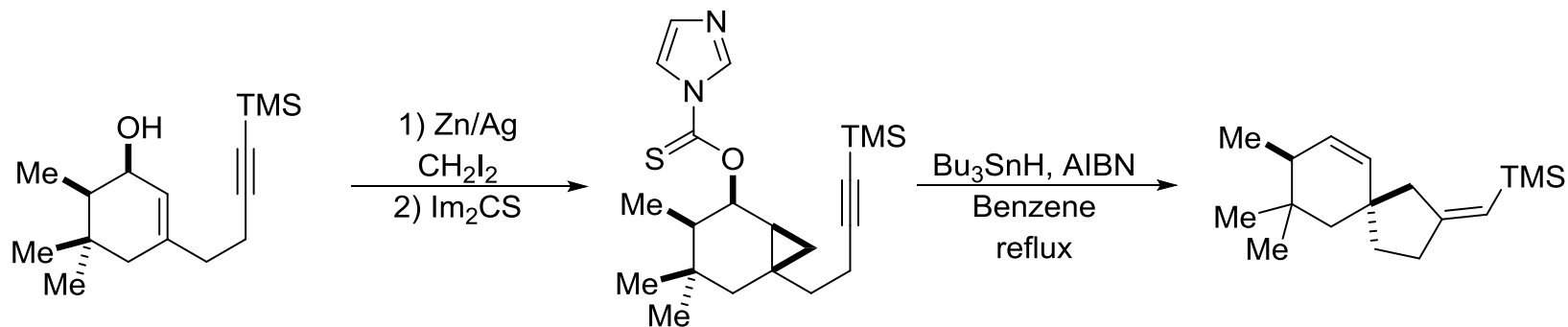
# 自由基引发的断裂反应

Rearrangements are possible

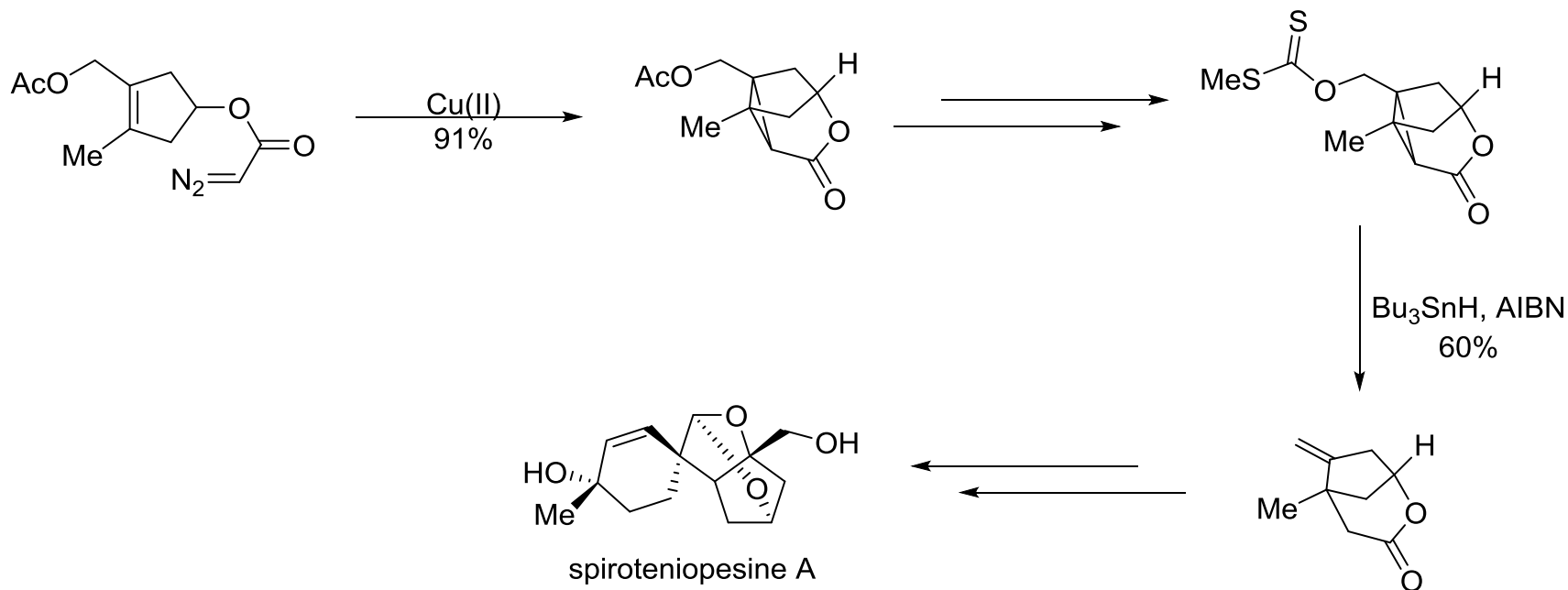


可用于检验反应中是否涉及自由基中间体

# 自由基引发的断裂反应的应用

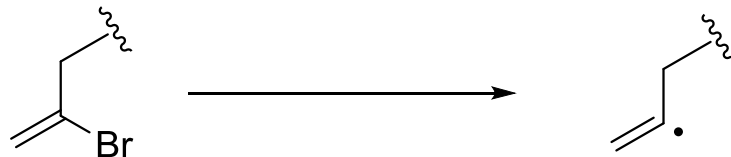


R. A. Batey, J. D. Harling, W. B. Motherwell, *Tetrahedron* **1992**, 48, 8031.

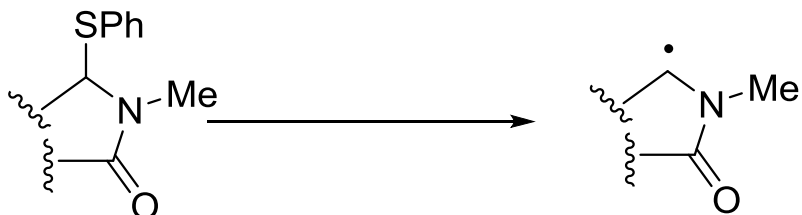


M. J. Dai, S. J. Danishefsky, *J. Am. Chem. Soc.* **2007**, 129, 3498.

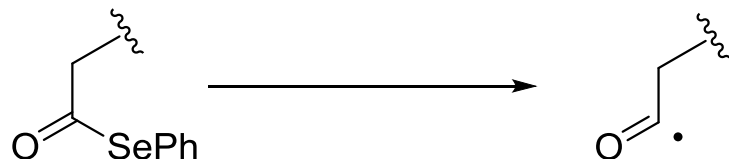
# 官能化的自由基



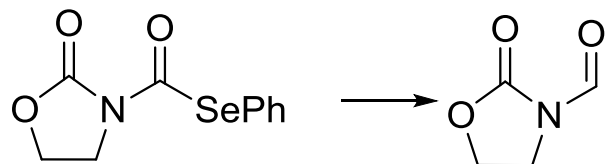
Stork, vinyl radicals



Hart. *J. Am. Chem. Soc.* **1997**, 117, 6226

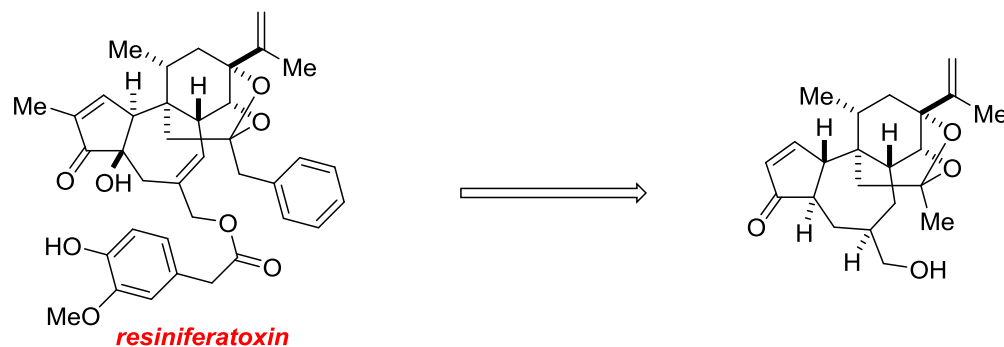
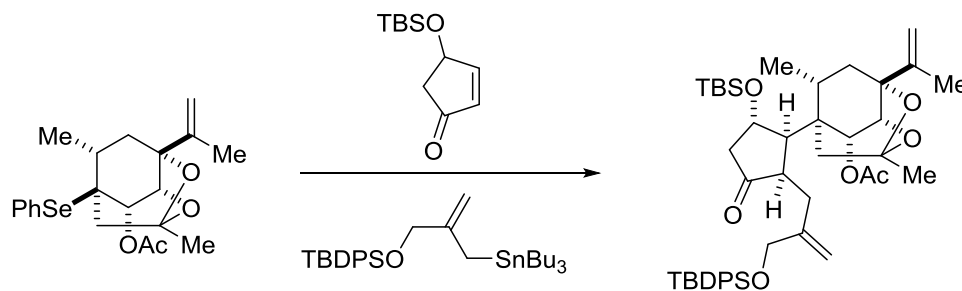
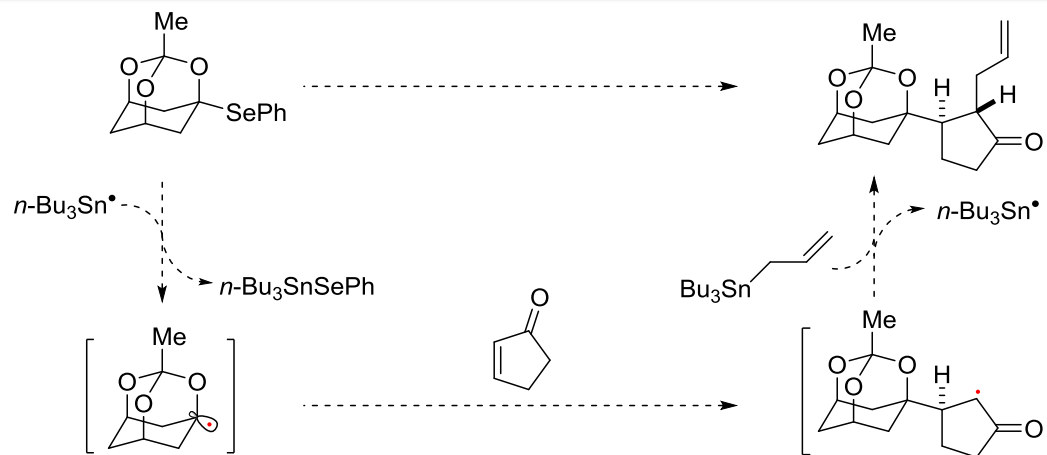


*Chem Rev*, **1999**, 99, 1991.



Keck, *Synlett*, **1999**, 1657

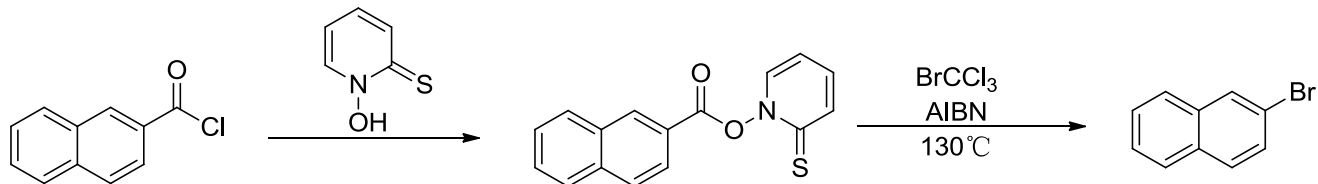
# 官能化的自由基



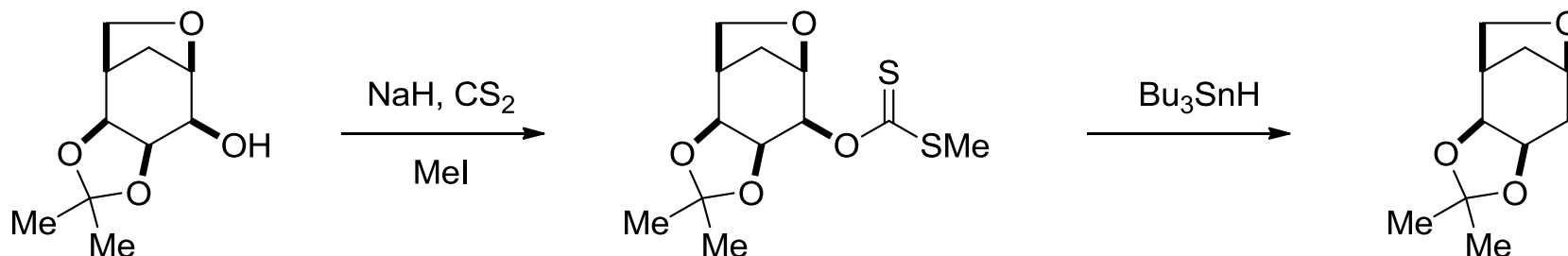
M. Inoue et al. *Chem. Sci.* **2013**, *4*, 2364.

# 几个重要的自由基人名反应

## Barton Decarboxylation

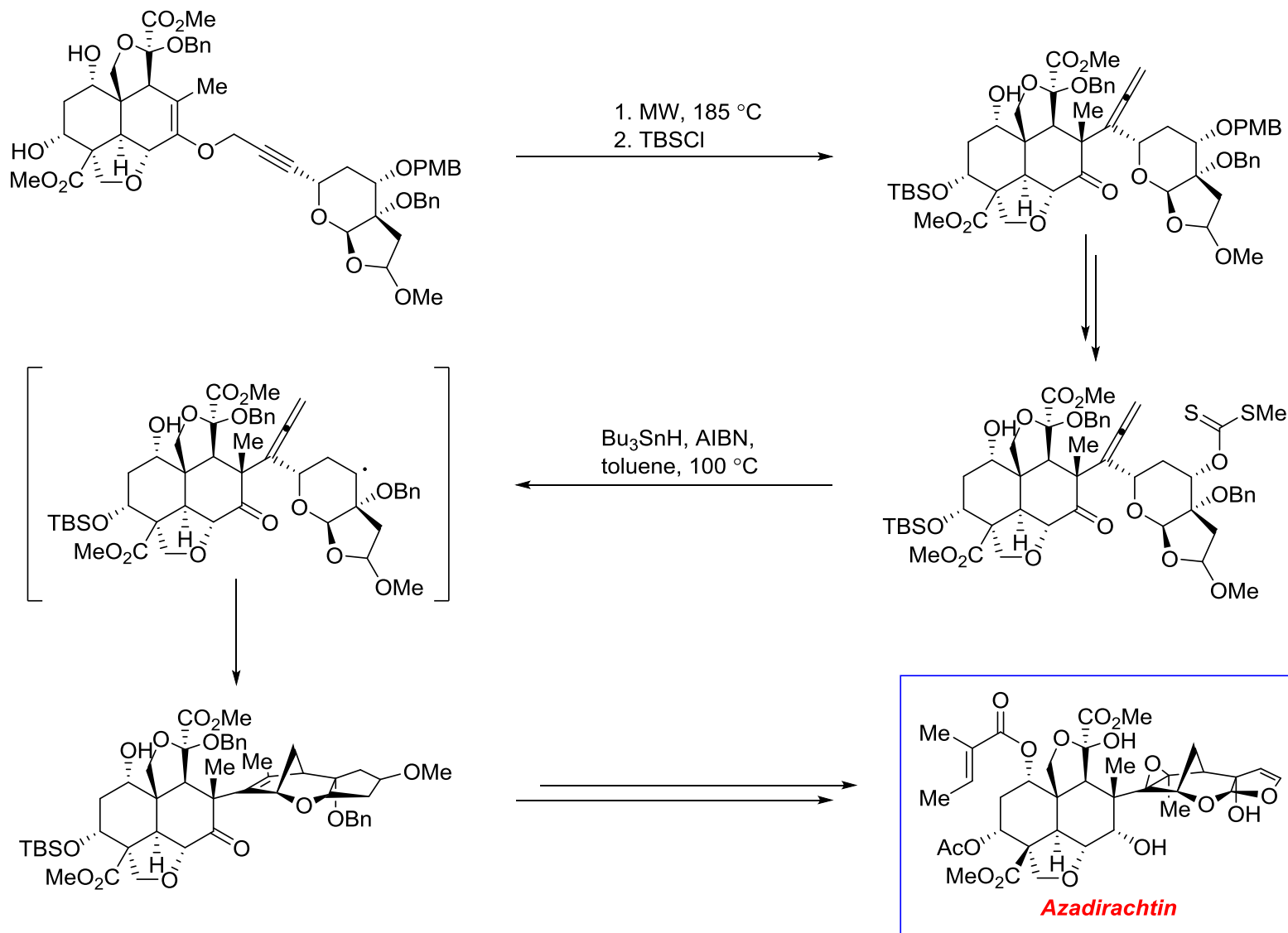


## Barton-McCombie Deoxygenation

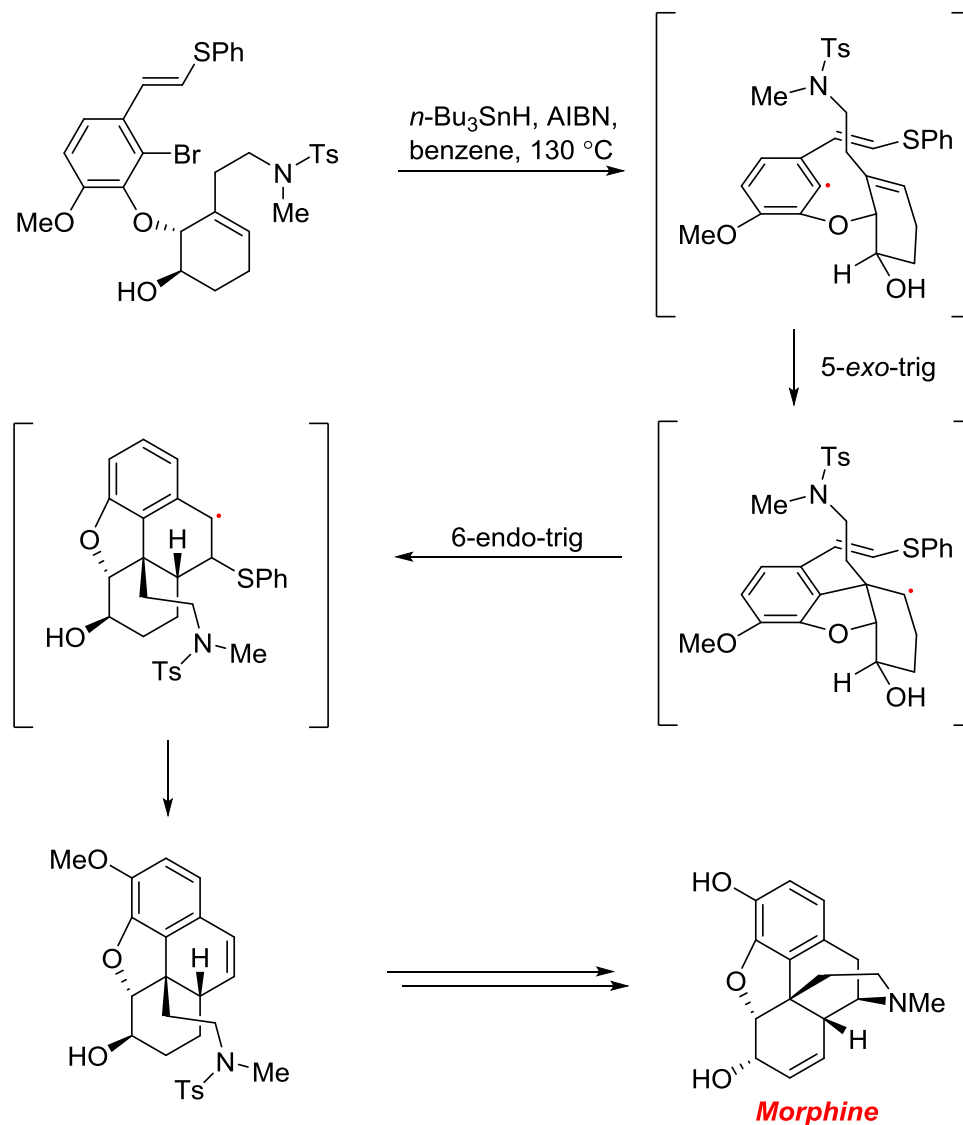


Barton, McCombie *J. Chem. Soc, Perkin1* **1975**,1574.

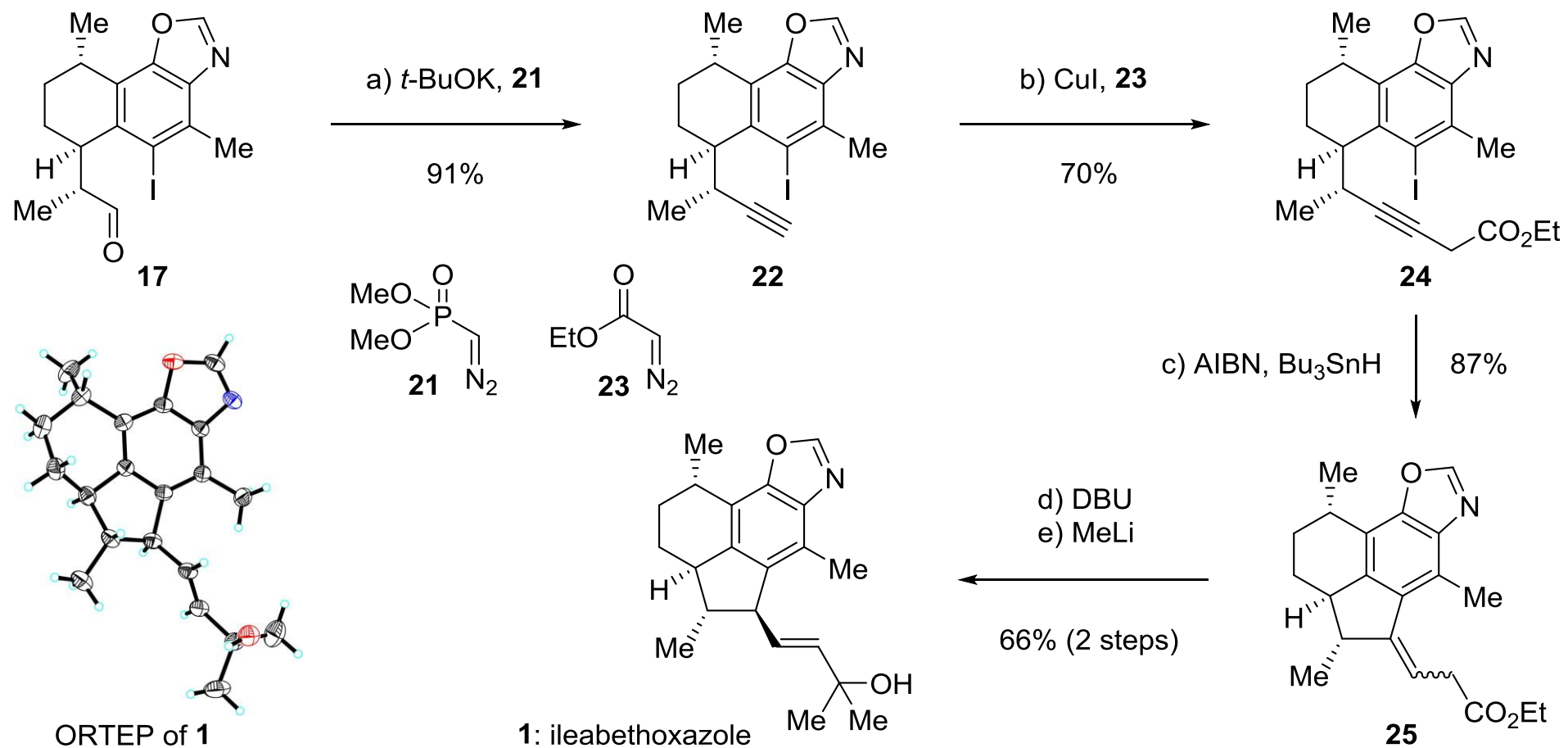
# 自由基参与的C-C键形成反应：实例1 桥环的构建



# 自由基参与的C-C键形成反应：实例4 季碳构建/串联反应



# 自由基参与的C-C键形成反应：实例5 自由基和其他反应的互补性



M. Yang, X. Yang, H. Sun, A. Li, *Angew. Chem. Int. Ed.* **2016**, 55, 2851.



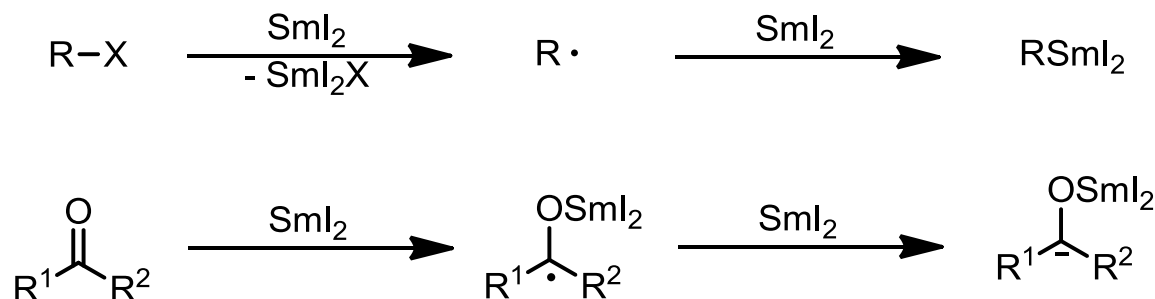
# Sml<sub>2</sub>的基本性质

制备: Sm/I<sub>2</sub>, Sm/ICH<sub>2</sub>CH<sub>2</sub>I

性质: reduction potential up to 2.05 V in the presence of HMPA

Aldrich: 0.1 M/100 mL, 625.95 RMB

a) Sml<sub>2</sub>-mediated activation of alkyl hldes

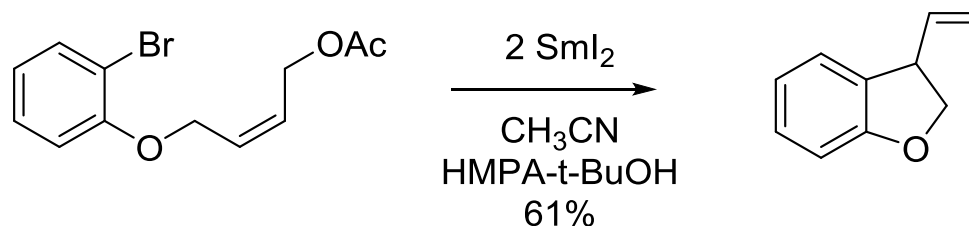


- Barbier reaction
- Ketyl-olefin coupling reaction
- Pinacol type coupling reaction
- Reduction

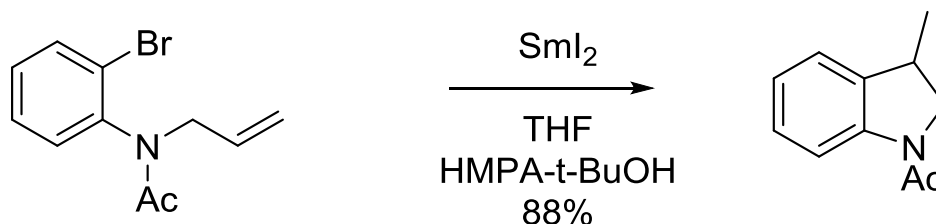
P. Girard, J. L. Namy and H. B. Kagan "Divalent Lanthanide Derivatives in Organic Synthesis. 1. Mild Preparation of Sml<sub>2</sub> and Ybl<sub>2</sub> and Their Use as Reducing or Coupling Agents". *J. Am. Chem. Soc.* **1980**, 102, 8, 2693.

K. C. Nicolaou, S. P. Ellery, J. S. Chen. "Samarium Diiodide Mediated Reactions in Total Synthesis". *Angew. Chem. Int. Ed.* 2009, **48**, 7140.

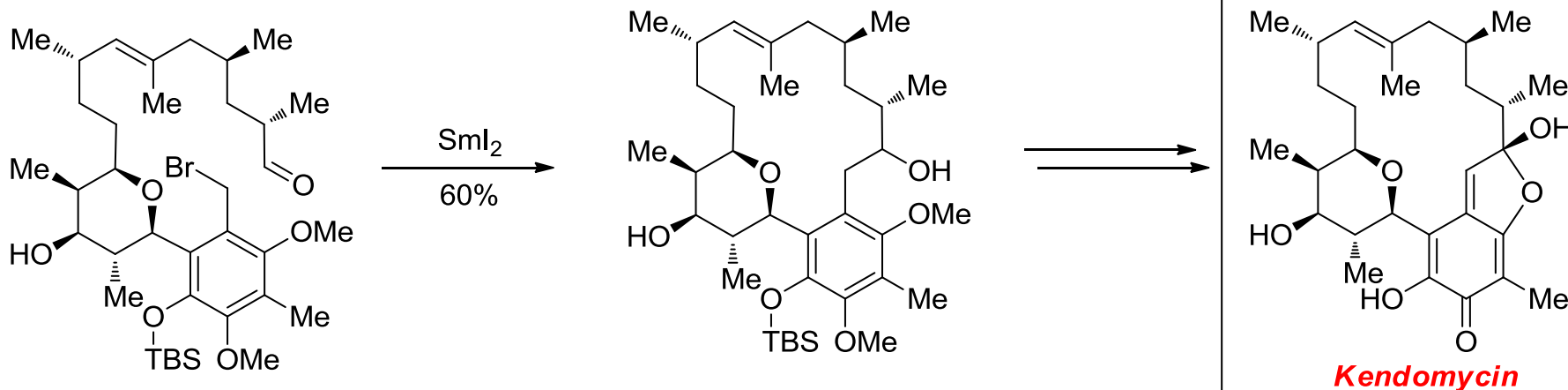
# $\text{Sml}_2$ 参与的C-C键形成反应：实例 Barbier reaction



Inanaga *Tetrahedron Lett.* **1991**, 32, 1737.



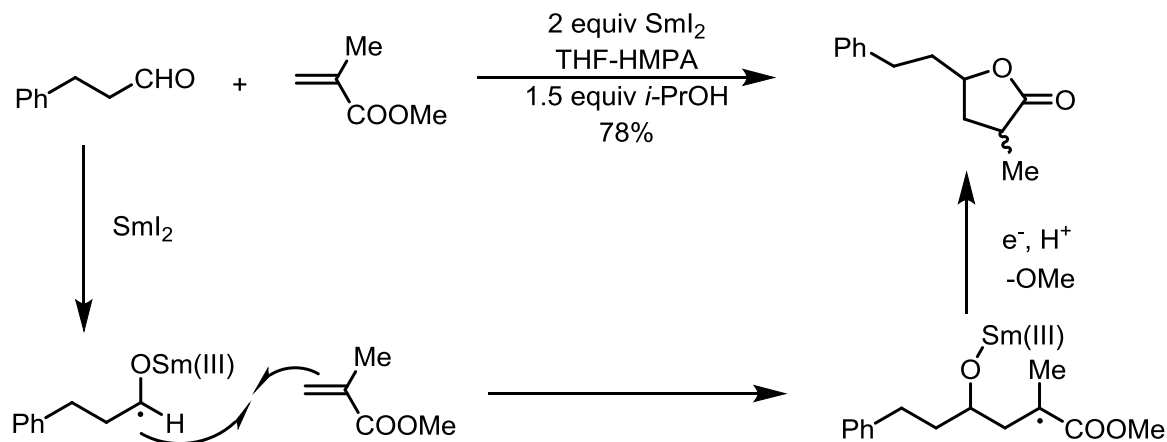
Molander *J.Org.Chem.* **1990**, 55, 6171.



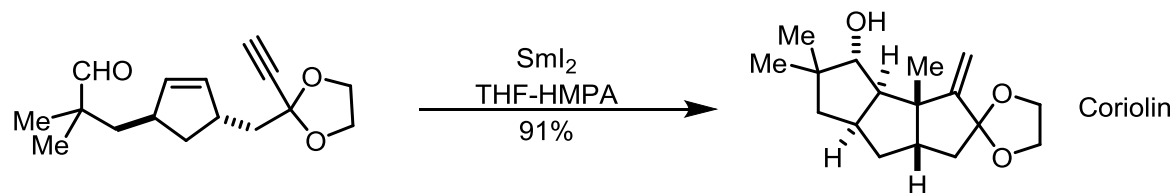
J. T. Lowe, J. S. Panek, *Org. Lett.* **2008**, 10, 3813.

**Kendomycin**

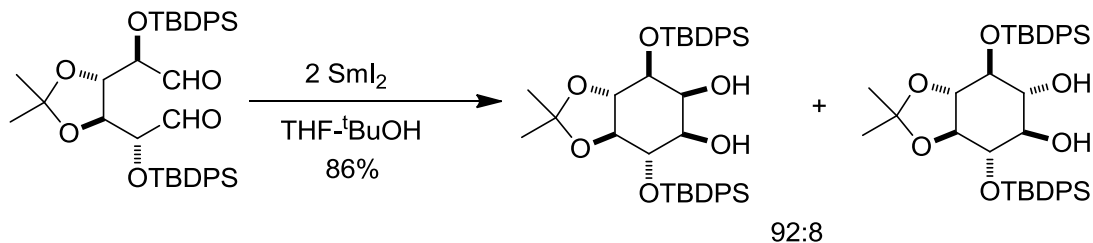
# Sml<sub>2</sub>参与的C-C键形成反应：羰基-烯烃和pinacol类型偶联



Inanaga, *Tetrahedron Lett.* **1986**, 27, 5763

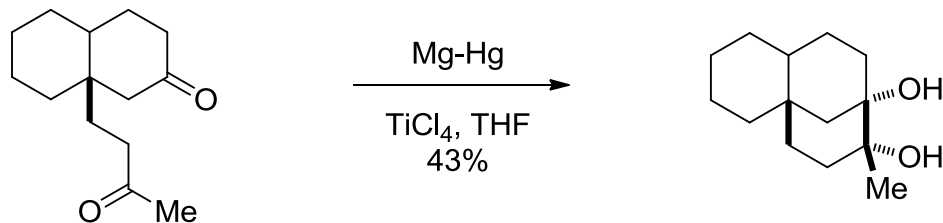


Curran, *J. Am. Chem. Soc.* **1988**, 110, 5064.

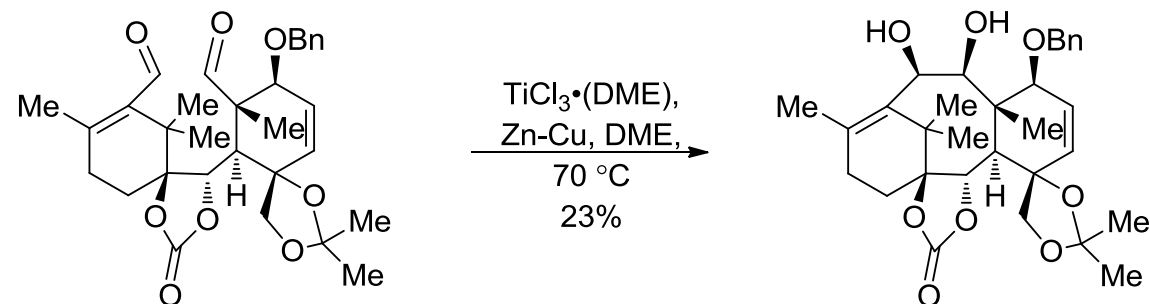


Chiara *Tetrahedron Lett.* **1994**, 35, 2969.

# Ti(III)参与的C-C键形成反应：McMurry反应



McMurry *J Org Chem.* **1977**, 42, 2655.



K. C. Nicolaou, Z. Yang, et al. *Nature* **1994**, 367, 630.

